

## Side-Chain Dynamics of Two Aromatic Amino Acids in Pancreatic Phospholipase A<sub>2</sub> As Studied by Deuterium Nuclear Magnetic Resonance<sup>†</sup>

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**ABSTRACT:** The flexibility of individual amino acid side chains of pancreatic phospholipase A<sub>2</sub> in aqueous and micellar solutions was studied with deuterium nuclear magnetic resonance (<sup>2</sup>H NMR). Bovine pancreatic phospholipase A<sub>2</sub> was selectively deuterated at the aromatic ring systems of Trp-3 and Phe-5 and porcine pancreatic phospholipase A<sub>2</sub> at Trp-3 only. Solid-state <sup>2</sup>H NMR spectra of the lyophilized enzymes exhibited quadrupole splittings on the order of 130 kHz, indicating almost complete immobilization of the aromatic ring systems. Exposure to a water-saturated atmosphere did not remove these steric constraints. However, side-chain mobility could be induced for the tryptophyl residue of the bovine enzyme by dissolving this enzyme in aqueous buffer or micellar solution whereas the phenyl ring always remained immobile and served as a probe for the protein's overall rotation. Typical correlation times for the tryptophyl and phenyl aromatic ring systems in aqueous solution were 7 ps and 13 ns (at 20 °C), respectively. The correlation time of the phenyl ring was longer than expected for the monomeric protein (~6 ns), suggesting some aggregation of the protein at the high concentrations used for the NMR measurements. Addition of a micellar solution of oleoylphosphocholine had no influence on the motional freedom of the tryptophyl residue but approximately doubled the correlation time of the phenyl ring, indicating an increase of the effective volume of the tumbling particle due to lipid-protein interaction. A different behavior was observed for the Trp-3 residue of porcine phospholipase A<sub>2</sub>. In aqueous solution, this residue was characterized by a broad resonance with a correlation time of about 13 ns (at 20 °C) which indicated immobilization of this residue in the protein structure and also partial aggregation of the enzyme. Upon addition of oleoylphosphocholine micelles, the <sup>2</sup>H NMR spectrum completely disappeared. This is explained by the formation of a large protein-lipid complex which broadens the <sup>2</sup>H NMR line width beyond detectability.

The biological membrane is a mixture of lipids and proteins, and knowledge of the dynamic properties of both components is of considerable interest in developing models for membrane function. Nuclear magnetic resonance (NMR)<sup>1</sup> provides a most general experimental technique to observe a broad range of molecular motions, and <sup>2</sup>H NMR in particular has led to a distinct improvement in our knowledge of lipid dynamics in bilayer membranes [for reviews, cf. Seelig & Seelig (1980) and Davis (1983)]. Using selectively deuterated lipids, <sup>2</sup>H NMR has yielded a detailed description of the lipid conformation and the rates of hydrocarbon chain and polar head-group motions. By employing membrane reconstitution procedures, it has also been possible to monitor the influence of integral membrane proteins on the lipid conformation (Seelig et al., 1981, 1982; Tamm & Seelig, 1983; Rice et al., 1979; Paddy et al., 1981).

In the present study, we have addressed the complementary question of how the internal flexibility of a protein can be influenced by interaction with an organized lipid-water interface. This problem was approached by selectively deuterating phospholipase A<sub>2</sub> and monitoring the dynamic behavior of the amino acid side chains in the presence and absence of lipids.

The catalytic activity of phospholipase A<sub>2</sub> consists in cleaving off the *sn*-2 fatty acyl chain of phospholipids. Characteristic features of extracellular phospholipases from pancreatic juice and snake or insect venom are their low molecular weight (~14 000), a requirement for Ca<sup>2+</sup> ions for activity, and a high stability under denaturing conditions. The latter may be explained by the presence of seven disulfide bridges [for a review, cf. Slotboom et al. (1982)]. Furthermore, and most important for NMR studies, pancreatic enzymes can be purified in large quantities.

The complete amino acid sequences of both the porcine (Puijk et al., 1977) and bovine (Fleer et al., 1978) enzymes are known. The two enzymes show a high homology in primary structure. They are secreted as almost inactive pro-enzymes which are activated by limited trypsinolysis. The activation of the phospholipases is accompanied by a conformational change in the N-terminal region of the protein.

An important step toward a detailed description of the mechanism of action of pancreatic phospholipase A<sub>2</sub> has been determination of the three-dimensional structure of the bovine enzyme by X-ray diffraction analysis with a resolution of 1.7 Å (Dijkstra et al., 1981a; Verheij et al., 1980). The molecule has the shape of a flattened cylinder with dimensions of approximately 22 × 30 × 42 Å<sup>3</sup>. About 50% of the amino acids are in the α-helix conformation and about 10% in the β-structure. Recently, the X-ray structure of the porcine enzyme has also been solved (Dijkstra et al., 1983).

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<sup>1</sup> Abbreviations: IRS, interface recognition site; NMR, nuclear magnetic resonance; AMPA, fully ε-amidated phospholipase A<sub>2</sub>.

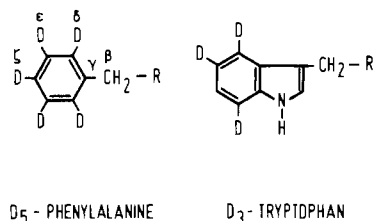


FIGURE 1: Deuteration sites of amino acid side chains Trp-3 and Phe-5.

The activity of pancreatic phospholipase A<sub>2</sub> is highly dependent on the physical state of the substrate. The observation that the hydrolysis rate of the enzyme is distinctly increased if the substrate is present in an aggregated form has led to the hypothesis of a specific interface recognition site (IRS), separate from the active site (Verger et al., 1973; Pierson et al., 1974; Brockerhoff, 1973). On the basis of the X-ray structural analysis, Dijkstra et al. (1981b) described the interfacial recognition site of bovine phospholipase A<sub>2</sub> in terms of three subdomains containing mainly hydrophobic amino acids of the N-terminal region, which is consistent with earlier spectroscopic evidence (van Dam-Mieras et al., 1975). One of these amino acids residues, namely, Trp-3, was therefore specifically deuterated in this study.

The other amino acid residue investigated, Phe-5, is located in the same  $\alpha$ -helical structure but directed toward the active site in the interior of the protein. The functional role of Phe-5 is not clear yet, but since it is part of the hydrophobic wall surrounding the active center, it may well be involved in the binding of the substrate monomer. Substitution of Phe-5 by Tyr in bovine phospholipase A<sub>2</sub> destroys the active site but does not affect the lipid binding properties (van Scharrenburg et al., 1982).

Using <sup>2</sup>H NMR, we have studied the protein mobility and side-chain flexibility of deuterated bovine and porcine phospholipases in the lyophilized state, in aqueous solution, and in the presence of micellar substrates.

#### MATERIALS AND METHODS

Phenylalanine which was perdeuterated at the aromatic ring system (Figure 1) was purchased from Merck Sharp & Dohme, Canada. Bovine and porcine pancreatic phospholipases A<sub>2</sub> were purified from pancreatic tissue and converted into the corresponding phospholipases A<sub>2</sub> by limited proteolysis as described by Fleer et al. (1978) and Nieuwenhuizen et al. (1974), respectively. Bovine phospholipase A<sub>2</sub> concentrations were calculated from the absorbance at 280 nm with an  $E_{1\text{cm}}^{1\%}$  value of 12.5. Enzymatic activities were routinely determined by the titrimetric assay procedure using egg yolk lipoproteins as substrates (Nieuwenhuizen et al., 1974). In contrast to the described procedure, a 2-fold higher Ca<sup>2+</sup> concentration and a 1.5-fold higher sodium deoxycholate concentration were used. *cis*-9-Octadecylphosphocholine was synthesized as described by van Dam-Mieras et al. (1975).

**Bovine Phe<sup>5-2</sup>H<sub>5</sub>-Labeled  $\epsilon$ -Amidinated Phospholipase A<sub>2</sub>.** A phenylalanine deuterated at all phenyl ring positions was incorporated at position 5 of bovine  $\epsilon$ -amidinated phospholipase A<sub>2</sub> (AMPA) by a semisynthetic pathway (van Scharrenburg et al., 1981). The deuterated bovine AMPA was indistinguishable from native AMPA with respect to catalytic, substrate binding, and Ca<sup>2+</sup> binding properties. Bovine  $\epsilon$ -amidinated phospholipase A<sub>2</sub> possesses approximately 75% of the enzymatic activity of the native phospholipase A<sub>2</sub> on micellar dioctanoyllecithin, and the Ca<sup>2+</sup> and substrate binding properties of both enzymes are identical.

**Bovine and Porcine [Trp<sup>3-2</sup>H<sub>3</sub>]Phospholipases A<sub>2</sub>.** The lyophilized phospholipases A<sub>2</sub> were treated for 2 h under ni-

trogen at room temperature with CF<sub>3</sub>COOD (Bak et al., 1969; Holt et al., 1971). Subsequently, the enzymes were exhaustively dialyzed against 0.1% acetic acid in order to remove all possibly present deuterium-labeled small molecules. After lyophilization, the proteins were dissolved in a 5 mM acetate buffer, pH 6.0, and subjected for 5 min to a temperature of 80 °C in order to back-exchange the amide protons (Campbell et al., 1975). After this heat treatment, the phospholipases A<sub>2</sub> were purified on CM-cellulose as described before (Fleer et al., 1978; Nieuwenhuizen et al., 1974). The fractions containing the deuterated phospholipases A<sub>2</sub> were pooled, again exhaustively dialyzed against distilled H<sub>2</sub>O, and lyophilized. The purified enzymes possessed full enzymatic activities. A <sup>1</sup>H NMR (360-MHz) spectrum of bovine [Trp<sup>3-2</sup>H<sub>3</sub>]-phospholipase A<sub>2</sub> was obtained with a Bruker HX-360 spectrometer at the S.O.N.-Facility in Groningen. The aromatic region of the spectrum showed a decrease of approximately 70% of the intensities of the H-4, H-5, and H-7 proton resonances of the Trp-3 residue, proving the labeling of this unique Trp residue in phospholipase A<sub>2</sub>. The above described procedure has been shown to be specific for the labeling of exclusively Trp residues in various proteins (Bak et al., 1969; Holt et al., 1971). To unambiguously exclude the presence of any <sup>2</sup>H-labeled amino acid residues other than the unique Trp-3 in phospholipase A<sub>2</sub>, we applied the same labeling procedure to des(Ala<sup>1</sup>-Met<sup>8</sup>) bovine phospholipase A<sub>2</sub> (van Scharrenburg et al., 1981). This C-terminal protein fragment consists of 115 amino acid residues and lacks the N-terminal octapeptide H<sub>2</sub>N-Ala-Leu-Trp-Gln-Phe-Asn-Gly-Met. As had to be expected, the Trp lacking des(Ala<sup>1</sup>-Met<sup>8</sup>)phospholipase A<sub>2</sub> did not express any detectable <sup>2</sup>H NMR signal. Therefore, these findings convincingly demonstrate that in the phospholipases A<sub>2</sub> exclusively Trp<sup>3</sup> has been deuterated and that no deuterated small molecules were present.

**<sup>2</sup>H NMR Measurements.** NMR samples were prepared as follows. About 20 mg of deuterium-labeled phospholipase A<sub>2</sub> was dissolved in 0.5 mL of deuterium-depleted buffer (50 mM sodium acetate/acetic acid, pH 6.0, containing 100 mM NaCl) and filled either in a thick-walled test tube (8-mm o.d.) for T<sub>1</sub> measurements using a solenoid probe head or in a conventional NMR sample tube for recording high-resolution <sup>2</sup>H NMR spectra with a high-resolution probe head (Helmholtz coils). To study the lipid-protein interaction, 20 mg of oleoylphosphocholine and 7.4 mg of CaCl<sub>2</sub>·2H<sub>2</sub>O yielding a final CaCl<sub>2</sub> concentration of 0.1 M were added to the phospholipase sample. The critical micellar concentration of oleoylphosphocholine is 7  $\mu$ M (Donné-Op den Kelder et al., 1981).

All <sup>2</sup>H NMR measurements were made on a Bruker-Spectrospin CXP-300 spectrometer operating at 46.1-MHz deuterium frequency. The experimental conditions are similar to those described earlier (Tamm & Seelig, 1983). Solid-state spectra were recorded with a solenoid coil using the quadrupole echo technique (Davis et al., 1976); 90° pulses of 4–5- $\mu$ s duration were used with an echo separation of 30  $\mu$ s. T<sub>1</sub> measurements of aqueous solutions were made with the same probe head. The recovery of the magnetization after a 180° inverting pulse was followed by using the quadrupole echo. Since the solenoid probe head could not be shimmed better than about 100 Hz, all line-width measurements were made with a high-resolution probe head (no lock signal); 90° pulses of 25- $\mu$ s length were used in this case.

#### RESULTS

**<sup>2</sup>H NMR Studies on Solid Phospholipase A<sub>2</sub>.** Using the quadrupole echo technique (Davis et al., 1976) <sup>2</sup>H NMR

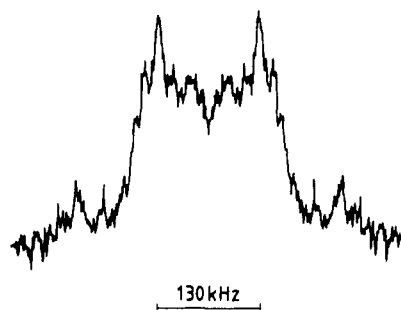


FIGURE 2: Solid-state  $^2\text{H}$  NMR spectrum of Trp-3-deuterated bovine pancreatic phospholipase  $\text{A}_2$ . Conditions: 85 mg of lyophilized enzyme measured at 25 °C; spectral width 500 kHz; recycle delay 5 s; 37 000 scans. The spectrum was measured with quadrature detection and symmetrized to improve the signal to noise ratio.

spectra can be recorded of peptides and proteins in the solid or semisolid state. The line shape of the solid-state  $^2\text{H}$  NMR spectrum is particularly sensitive to ring motions of aromatic amino acid side chains (Rice et al., 1981; Gall et al., 1981; Kinsey et al., 1981). Three extreme cases of ring motion can be discerned: (i) no ring motion at all, (ii) 2-fold jumps whereby the aromatic rings execute 180° reorientational flips, and (iii) rapid rotational ring diffusion. We were interested in which types of motions occur in lyophilized phospholipase  $\text{A}_2$ .

Figure 2 shows the  $^2\text{H}$  NMR spectrum of lyophilized bovine pancreatic phospholipase  $\text{A}_2$ , selectively deuterated at Trp-3. Since only 85 mg (6  $\mu\text{mol}$ ) of enzyme was available, the rather low signal to noise ratio of the powder pattern prohibits a detailed line-shape simulation. However, the separation of the most intense peaks in the spectrum can easily be measured and amounts to approximately 130 kHz. On the basis of a static quadrupole coupling constant of 183 kHz for the aromatic deuterons of tryptophan (Rice et al., 1981), a maximum separation of 137 kHz can be calculated for a completely rigid Trp-3 side chain. Hence, Figure 2 demonstrates an almost complete immobilization of the tryptophyl residue.  $^2\text{H}$  NMR spectra were also recorded for the Phe-5-labeled bovine enzyme and the Trp-3-labeled porcine enzyme. Both phospholipases yielded powder patterns with about 130-kHz separation, indicating that both ring systems were also immobilized in the lyophilized protein. This situation was not changed when both Trp-3-labeled enzymes were exposed to a water-saturated atmosphere for 24 h.

**Aqueous Solutions of Phospholipase  $\text{A}_2$ .** The pancreatic phospholipases are water-soluble enzymes. For a globular protein with a molecular weight of 14 000, the rotational diffusion constant can be predicted to be about  $D_r = 10^7\text{--}10^8 \text{ s}^{-1}$ , yielding a characteristic correlation time  $\tau_c = (6D_r)^{-1} = 20\text{--}2 \text{ ns}$ . Due to this rapid isotropic tumbling, the quadrupole splitting will be averaged out, and the  $^2\text{H}$  NMR spectrum will consist of a single resonance. However, dynamic information can still be extracted from measurements of the spin-lattice ( $T_1$ ) relaxation time and the line width as has been demonstrated recently for lysozyme (Wooten & Cohen, 1979; Schramm & Oldfield, 1983). The main advantage of using  $^2\text{H}$  NMR in these experiments stems from the fact that the dominant relaxation pathway is quadrupolar relaxation, and the molecular interpretation of the  $T_1$  relaxation time and the line width is thus governed by rather simple formalisms.

The outcome of two typical  $T_1$  relaxation time measurements with bovine and porcine phospholipase  $\text{A}_2$  labeled at the same amino acid, Trp-3, is displayed in Figure 3. The bovine enzyme is characterized by a sharp line with a long  $T_1$  relaxation time ( $T_1 = 300 \text{ ms}$  at 20 °C) whereas the porcine

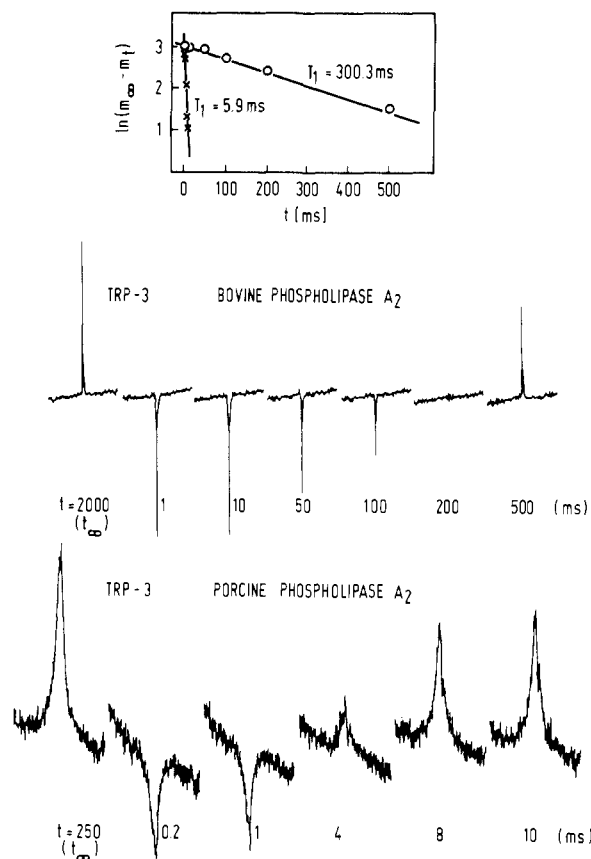


FIGURE 3: Measurement of deuterium spin-lattice ( $T_1$ ) relaxation time of Trp-3-labeled bovine (O) and porcine (X) phospholipase  $\text{A}_2$  in aqueous solution. Approximately 20 mg of phospholipase was dissolved in 0.5 mL of deuterium-depleted buffer. Conditions: spectral width 10 kHz; 1000 scans per spectrum. Spectra were measured at 20 and 25 °C for the bovine and porcine enzymes, respectively.

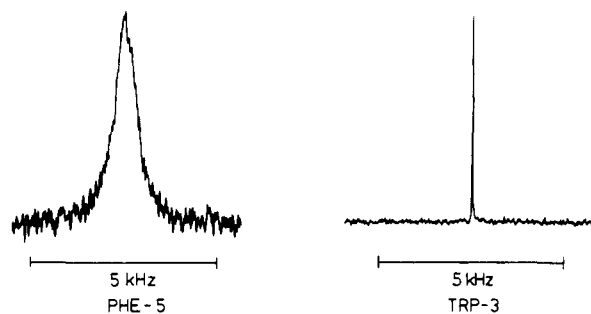


FIGURE 4: High-resolution  $^2\text{H}$  NMR spectra of aqueous solutions of bovine phospholipase  $\text{A}_2$  deuterated at Trp-3 and Phe-5, respectively. Measuring temperature, 25 °C. Forty milligrams of the protein was dissolved in 1 mL of deuterium-depleted buffer; 2500 and 10 000 scans were recorded for the Trp-3- and Phe-5-deuterated proteins, respectively.

enzyme gives rise to a broad line with a short  $T_1$  ( $T_1 = 5.9 \text{ ms}$  at 25 °C).

A comparison of two different amino acids, Trp-3 and Phe-5, within the same phospholipase  $\text{A}_2$  (bovine) is reproduced in Figure 4. Again one notes a considerable difference in line width between the two aromatic residues, indicating distinct motional differences. Spectral simulations were carried out for the broad resonances (assuming Lorentzian line shapes) in order to obtain a more precise estimate of the line width. Table I provides a summary of the line widths and  $T_1$  relaxation time data of the various amino acid residues investigated.

The interaction of phospholipase  $\text{A}_2$  with a lipid substrate was investigated by adding oleoylphosphocholine and  $\text{CaCl}_2$

in solid form to the aqueous solution of the enzyme. The resulting solution was characterized by the following concentrations: phospholipase A<sub>2</sub>, 2.9 mM; oleoylphosphocholine, 89.5 mM; and CaCl<sub>2</sub>, 100 mM. In the presence of Ca<sup>2+</sup>, the enzyme binds to the micellar surface as has been demonstrated previously for various *n*-alkylphosphocholine analogues (de Araujo et al., 1979; Hille et al., 1981; Donn -Op den Kelder, 1981). Table I shows a comparison of the *T*<sub>1</sub> relaxation times and the line widths with and without micelles as a function of temperature. No general conclusions can be drawn from Table I since the addition of micelles produces different effects for the three residues investigated. Trp-3 bovine phospholipase A<sub>2</sub> exhibits practically no difference between aqueous and micellar solution, while the Phe-5 residue of the same enzyme experiences an almost 2-fold increase in the *T*<sub>1</sub> relaxation time and also an increase in line width. Finally, the Trp-3 resonance of the porcine enzyme is broadened beyond detection in the presence of micelles. The latter process is reversible since the <sup>2</sup>H NMR signal can be recovered by removing the micelles with gel exclusion chromatography (Slotboom et al., 1977).

## DISCUSSION

<sup>2</sup>H NMR is well suited for studying the mobility in a protein. A prerequisite for obtaining meaningful results is the synthesis of a selectively deuterated protein which was achieved here for two different aromatic amino acid side chains.

The measurements with the lyophilized phospholipase A<sub>2</sub> demonstrate that 6 μmol of enzyme is sufficient to yield <sup>2</sup>H NMR solid-state spectra of reasonable quality. All three enzymes investigated exhibit powder patterns with a 130-kHz splitting, indicating an almost complete immobilization of the aromatic ring systems. Furthermore, it was not possible to induce the rotation of the Trp-3 residue by exposing the phospholipase A<sub>2</sub> to a water-saturated atmosphere even though the same residue is freely rotating in the dissolved enzyme, at least for the bovine enzyme.

If one assumes isotropic reorientation characterized by a single correlation time *τ*<sub>c</sub>, then, as is well established (Abragam, 1961)

$$\frac{1}{T_1} = \frac{3}{20}\pi^2 \left[ \frac{e^2 q Q}{h} \right]^2 \left[ 1 + \frac{\eta^2}{3} \right] \left[ \frac{2\tau_c}{1 + \omega_0^2 \tau_c^2} + \frac{8\tau_c}{1 + 4\omega_0^2 \tau_c^2} \right] \quad (1)$$

and

$$\Delta\nu_{1/2} = \frac{1}{\pi T_2} = \frac{3}{20}\pi \left[ \frac{e^2 q Q}{h} \right]^2 \left[ 1 + \frac{\eta^2}{3} \right] \times \left[ 3\tau_c + \frac{5\tau_c}{1 + \omega_0^2 \tau_c^2} + \frac{2\tau_c}{1 + 4\omega_0^2 \tau_c^2} \right] \quad (2)$$

where *η* is the asymmetry parameter, *ω*<sub>0</sub> the Larmor precessional frequency, and *Δν*<sub>1/2</sub> the line width at half-height. The spin-lattice relaxation curve exhibits a minimum around *τ*<sub>c</sub> ~ *ω*<sub>0</sub><sup>-1</sup>, and a single *T*<sub>1</sub> measurement is usually not sufficient to allow an unambiguous determination of *τ*<sub>c</sub>. However, in combination with the line width or the measurement of the temperature dependence of *T*<sub>1</sub>, the assignment of a unique *τ*<sub>c</sub> value becomes straightforward.

**Bovine Phospholipase A<sub>2</sub>.** Inspection of Table I reveals that Phe-5-deuterated bovine phospholipase A<sub>2</sub> exhibits a negative temperature dependence in *T*<sub>1</sub>; i.e., increasing temperatures lead to shorter *T*<sub>1</sub> relaxation times. This behavior is typical

for the so-called slow correlation time regime where *ω*<sub>0</sub> > *τ*<sub>c</sub><sup>-1</sup>, and eq 1 predicts correlation times (*τ*<sub>c</sub>) in the range of 12–25 ns, assuming a quadrupole coupling constant of 180 kHz and neglecting the asymmetry parameter (*η* ≈ 0.03; Rice et al., 1981). Insertion of the calculated *τ*<sub>c</sub> values into the line-width eq 2 yields a very good agreement with the experimentally observed line widths listed in Table I.

The correlation time *τ*<sub>c</sub> is related to the effective volume *V*<sub>eff</sub> of the reorienting sphere according to

$$\tau_c = \frac{V_{\text{eff}} \eta_0}{kT} \quad (3)$$

where *η*<sub>0</sub> is the viscosity of the solution, *k* the Boltzmann constant, and *T* the absolute temperature. The viscosity of the phospholipase A<sub>2</sub> solution was not measured directly but can be estimated by comparison with solutions of ribonuclease A which has a similar molecular weight [cf. Buzzell & Tanford (1956)]. For a phospholipase concentration of 40 mg/mL, which is typical for the NMR experiments, we estimate a viscosity of *η*<sub>0</sub> = 1.3 cP (at 20 °C), which together with *τ*<sub>c</sub> = 13.5 ns yields an effective volume of 4.2 × 10<sup>-20</sup> cm<sup>3</sup>. On the basis of the crystallographic dimensions, the total volume of the protein is predicted to be 2.8 × 10<sup>-20</sup> cm<sup>3</sup>. As a first conclusion, it therefore follows that the phenyl ring is embedded rigidly in the protein and acts as a probe for the overall protein rotation.

The effective volume as determined by NMR is larger than that calculated from the crystallographic data. This difference may be explained by the hydration of the enzyme in solution, by the uncertainty in the estimated viscosity, and perhaps by a partial self-association of phospholipase A<sub>2</sub>. It is noteworthy in this respect that lysozyme, an enzyme of similar molecular weight (13 900), exhibits correlation times from 6 ns (monomeric state) to about 23 ns depending on the state of aggregation of the enzyme [Wooten & Cohen, 1979; cf. also p 479 of Jardetzky & Roberts (1981)].

A completely different behavior is observed for the Trp-3 residue of bovine phospholipase. In solution, this residue gives rise to a rather narrow line of less than 5 Hz line width which together with the positive temperature dependence of *T*<sub>1</sub> indicates a motional behavior typical for the fast correlation time regime (*ω*<sub>0</sub> << *τ*<sub>c</sub><sup>-1</sup>). The correlation time *τ*<sub>c</sub> is approximately 10 ps which is 3 orders of magnitude shorter than the correlation time of the Phe-5 ring. Similar short correlation times are obtained for small molecules in aqueous solution [cf. Mantsch et al. (1977) and Jardetzky & Roberts (1981)]. Hence, as a second result, it follows that the Trp-3 ring is moving freely in solution and is essentially unaffected by the attachment of the amino acid to the phospholipase polypeptide backbone.

We may now discuss the properties of bovine phospholipase A<sub>2</sub> in micellar solutions of oleoylphosphocholine. As indicated in Table I, the correlation time *τ*<sub>c</sub> of the Phe-5 ring increases by almost a factor of 2 compared to that of the aqueous solution. This means not only that the phenyl ring remains immobilized but also that the effective volume of the tumbling molecule is considerably increased due to the binding of lipid molecules to the protein. The doubling of the effective volume can be explained quantitatively by assuming that each protein binds 30–50 molecules of oleoylphosphocholine (*M*<sub>r</sub> 447).

This conclusion is consistent with earlier physicochemical studies on porcine pancreatic phospholipase A<sub>2</sub>. According to Donn -Op den Kelder et al. (1981), the porcine phospholipase A<sub>2</sub> forms complexes with saturated *n*-alkylphosphocholines which are generally composed of 2 protein molecules and 60–100 lipid monomers [cf. also de Araujo et al. (1979)

Table I: Spin-Lattice Relaxation Time ( $T_1$ ), Line Width at Half-Height ( $\Delta\nu_{1/2}$ ), and Correlation Time ( $\tau_c$ ) of Phospholipase A<sub>2</sub> Solutions

		temp (°C)	$T_1$ (ms) <sup>a</sup>	$\Delta\nu_{1/2}$ (Hz) <sup>b</sup>	$\tau_c$ (ns) <sup>c</sup>	$E_a$ (kJ/mol) <sup>d</sup>
Phe-5 bovine phospholipase A <sub>2</sub>	aqueous solution	0	10.3	1200	23.3	14.0
		10	7.2	790	16.1	
		20	6.1	670	13.5	
		25		630	12	
	micellar solution	10	14.4		32.7	7.1
		20	13.7		31.2	
		30	11.3		25.6	
		40		1200	25.3	
		50		1100	23.0	
Trp-3 bovine phospholipase A <sub>2</sub>	aqueous solution	0	170		0.0119	21.7
		10	203	<5	0.0099	
		20	290		0.0070	
		30	432		0.0047	
	micellar solution	0	164		0.0123	20.8
		10	263	<5	0.0076	
		20	304		0.0066	
Trp-3 porcine phospholipase A <sub>2</sub>	aqueous solution	5	6.7		15.4	
		15	5.6		12.7	
		25	5.9	650	13.4	
	micellar solution <sup>e</sup>					

<sup>a</sup>  $T_1$  measurements were made with a solenoid probe head and 4.5- $\mu$ s (90°) pulses. The intrinsic field inhomogeneity was about 100–200 Hz.

<sup>b</sup> Line width at half-height. Line-width measurements were made with a high-resolution probe head, but without a lock signal. As tested with <sup>2</sup>H<sub>2</sub>O samples, the line width was less than 2 Hz. <sup>c</sup> Correlation times were evaluated from eq 1 and 2. Quadrupole coupling constants of 180 and 183 kHz were assumed for the aromatic deuterons of the phenylalanine and the tryptophan residue, respectively. The asymmetry parameter was neglected since  $\eta < 0.05$  and enters only as  $\eta^2$  in the relaxation formulas. <sup>d</sup> Activation energies ( $E_a$ ) were determined from an Arrhenius plot of the correlation times ( $\tau_c$ ). <sup>e</sup> Resonance too broad; no signal observable.

and Hille et al. (1981)]. As the only exception, oleoyl-phosphocholine was found to form lipid-protein aggregates consisting of 3 porcine phospholipase A<sub>2</sub> molecules and 100 lipid monomers. However, bovine phospholipase A<sub>2</sub> may have different lipid binding properties than the porcine enzyme, and a complex of 2 bovine phospholipases A<sub>2</sub> and 60–100 lipid monomers of oleoylphosphocholine is in better agreement with the <sup>2</sup>H NMR results.

The correlation times determined by <sup>2</sup>H NMR furthermore indicate that the individual monomers are not interacting strongly but appear to rotate independently with their respective lipid shells. The correlation times of the whole complex can be estimated to be at least a factor of 2 larger than the measured correlation times.

The  $T_1$  relaxation time of the Trp-3 residue is not altered upon binding of phospholipase A<sub>2</sub> to micelles. The correlation time amounts to about 12–7 ps with and without micelles which suggests that the tryptophan aromatic ring system of the bovine enzyme remains in the water phase and rotates freely without any additional restrictions.

**Porcine Phospholipase A<sub>2</sub>.** Only the Trp-3, but not the Phe-5, ring was deuterated in this enzyme. However, the behavior of Trp-3 of porcine phospholipase A<sub>2</sub> is clearly different from that of the bovine enzyme (cf. Table I). In aqueous solution, the Trp-3 ring gives rise to broad resonances with  $\Delta\nu_{1/2} \approx 650$  Hz. From this and from the measurement of the spin-lattice relaxation  $T_1$ , a correlation time  $\tau_c = 12$ –15 ns can be derived which is again characteristic of the reorientation of the protein as a whole. Thus, in a 3 mM solution of the porcine enzyme, the Trp-3 ring is immobilized in the protein structure which may be contrasted with the free rotation of Trp-3 in the bovine enzyme under the same conditions. Moreover, upon addition of micelles, the Trp-3 resonance is no longer observable which suggests the formation of an even larger lipid-protein complex with a correlation time of  $\tau_c > 40$  ns.

In conclusion, the <sup>2</sup>H NMR studies thus have demonstrated that (1) there can be dramatic differences between the mobility of amino acid side chains, viz., Trp-3 and Phe-5, which are located on the same stretch of  $\alpha$ -helix, (2) the same residue

(Trp-3) can be immobilized or freely rotating depending on the origin of the phospholipase A<sub>2</sub>, and (3) the incorporation of phospholipase A<sub>2</sub> into micelles generally tends to reduce the overall rotation of the enzyme due to lipid binding and/or aggregation.

**Registry No.** Trp, 73-22-3; Phe, 63-91-2; phospholipase A<sub>2</sub>, 9001-84-7.

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## Intramolecular Electron and Proton Transfer in Proteins: $\text{CO}_2^-$ Reduction of Riboflavin Binding Protein and Ribonuclease A

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**ABSTRACT:** The formate radical ( $\text{CO}_2^-$ ) reacts with ribonuclease A to form the cystine disulfide radical as one of the products.  $\text{CO}_2^-$  reacts with the riboflavin binding protein of chicken egg white with the ultimate product being the neutral flavin semiquinone. Formation of the disulfide radical in ribonuclease is slower than the reaction between protein and  $\text{CO}_2^-$ ; formation of the flavin semiquinone in the riboflavin binding protein is slower than the protein- $\text{CO}_2^-$  reaction. We conclude for both proteins that  $\text{CO}_2^-$  must reduce an as yet unidentified group or groups, which in turn reduce(s) the disulfide of RNase or the flavin of riboflavin binding protein. This conclusion is supported in the case of ribonuclease by the observation of a transient, broad absorption band centered between 350 and 370 nm. The  $\text{CO}_2^-$ -initiated reductions of the disulfide in ribonuclease and the flavin in the riboflavin binding protein are mixed first- and second-order processes. We propose that the transfer of an electron from the unknown intermediate(s) to the final product involves both inter- and intramolecular paths between groups that may not be in van der Waals contact. With the hydrated electron, in contrast to  $\text{CO}_2^-$ , as reductant of the riboflavin binding protein, the anionic semiquinone is observed as an intermediate. The anionic semiquinone is then rapidly protonated, yielding the stable neutral semiquinone. From the reaction kinetics and protein concentration dependence, we conclude that a group or groups on the protein donate(s) a proton to the anionic semiquinone by both inter- and intramolecular paths.

**T**he proposal that electrons can migrate or transfer over long distances in proteins has been made a number of times, and for a variety of reasons [e.g., Winfield (1965), Grossweiner (1976), and Klapper & Faraggi (1979)]. Not only does the existence of such transfer have significance for the understanding of physiological redox reactions, but we have also suggested (Steiner et al., 1985) that the detection and study

of protein intramolecular electron transfer may yield information about internal electrostatic properties of proteins. There have been a number of recent reports on electron transfer between groups that are not in van der Waals contact in small model systems [e.g., Prutz et al. (1981), Guar et al. (1983), Calcaterra et al. (1983), and Isied & Vassilian (1984)]. Two groups (Winkler et al., 1982; Isied et al., 1984)